

ATP-driven stepwise rotation of F_0F_1 -ATP synthase

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F_0F_1 -ATP synthase (F_0F_1) is a motor enzyme that couples ATP synthesis/hydrolysis with a transmembrane proton translocation. F_1 , a water-soluble ATPase portion of F_0F_1 , rotates by repeating ATP-waiting dwell, 80° substep rotation, catalytic dwell, and 40° substep rotation. Compared with F_1 , rotation of F_0F_1 has yet been poorly understood, and, here, we analyzed ATP-driven rotations of F_0F_1 . Rotation was probed with an 80-nm bead attached to the ring of c subunits in the immobilized F_0F_1 and recorded with a submillisecond fast camera. The rotation rates at various ATP concentrations obeyed the curve defined by a K_m of ≈ 30 μM and a V_{max} of ≈ 350 revolutions per second (at 37°C). At low ATP, ATP-waiting dwell was seen and the $k_{\text{on-ATP}}$ was estimated to be 3.6×10^7 $\text{M}^{-1}\cdot\text{s}^{-1}$. At high ATP, fast, poorly defined stepwise motions were observed that probably reflect the catalytic dwells. When a slowly hydrolyzable substrate, adenosine 5'-[γ -thio]triphosphate, was used, the catalytic dwells consisting of two events were seen more clearly at the angular position of $\approx 80^\circ$. The rotational behavior of F_0F_1 resembles that of F_1 . This finding indicates that "friction" in F_0 motor is negligible during the ATP-driven rotation. Tributyltin chloride, a specific inhibitor of proton translocation, slowed the rotation rate by 96%. However, dwells at clearly defined angular positions were not observed under these conditions, indicating that inhibition by tributyltin chloride is complex.

ATP hydrolysis | binding change mechanism | membrane protein | single-molecule imaging

F_0F_1 -ATPase/synthase (F_0F_1) is a large protein complex (≈ 500 kDa) that catalyzes ATP synthesis/hydrolysis coupled with a transmembrane H^+ (proton)-translocation in bacteria, chloroplasts, and mitochondria (1–5). The enzyme is easily and reversibly separated into two portions, termed F_1 and F_0 . In its simplest prototype bacterial enzyme, a water-soluble F_1 portion consists of five different subunits, $\alpha_3\beta_3\gamma_1\delta_1\epsilon_1$, and catalyzes ATP hydrolysis (hence, often called F_1 -ATPase). Three α -subunits and three β -subunits are arranged alternately, forming a hexagonal cylinder around the coiled-coil structure of the γ -subunit (6). Membrane-integrated F_0 portion has three different subunits, $a_1b_2c_n$ (n ; variable among species) and mediates proton transport across the membrane. The c -subunits form a ring structure, and ab_2 associates with the c -subunit ring peripherally (7–10). F_0F_1 is a motor enzyme. When the magnitude of electrochemical potential of protons is large enough, downhill proton flow through F_0 causes rotation of the rotor subunits (c_n - $\gamma\epsilon$) relative to the stator subunits (ab_2 - $\alpha_3\beta_3\delta$), and rotation of the γ -subunit forces the β -subunits of F_1 to change conformations sequentially that result in ATP synthesis. In the reverse reaction, ATP hydrolysis at F_1 causes the reverse rotation of the rotor subunits that drives F_0 to pump protons (Fig. 1) (11).

We have been studying rotation of thermophilic *Bacillus* F_1 since the first direct visualization of ATP-driven rotation of F_1 immobilized on the glass surface (12). A rotation probe attached on the γ -subunit rotates unidirectionally counterclockwise when viewed from membrane side. It repeats a pause and a 120° step rotation when medium ATP concentration ([ATP]) is low (13). The duration of the pause becomes shorter as [ATP] increases and is finally invisible beyond the limit of the observation system. This [ATP]-dependent pause corresponds to the period during

which the enzyme waits for medium ATP to come into the empty catalytic site, and, hence, is called the ATP-waiting dwell. With high-speed imaging, it was found that the 120° step of rotation is further split into 90° and 30° substeps (14). A pause between two substeps, ≈ 2 ms at 23°C, is not influenced by [ATP]. This [ATP]-independent nature of the pause means that catalytic events after substrate binding should occur during this pause, and we call this pause the catalytic dwell. The histogram of durations of the catalytic dwells did not obey a single exponential but did obey double exponentials, indicating that two catalytic reactions of ≈ 1 ms occur in the catalytic dwell (14). More recent experiments using a slowly hydrolyzable ATP analog, adenosine 5'-[γ -thio]triphosphate (ATP γ S) and a slow mutant in ATP hydrolysis, clarified that one of the two 1-ms events is cleavage of a bound ATP at a catalytic site (15). Also, previous 90° and 30° substeps were recently revised to be 80° and 40° substeps (15, 16). Thus, F_1 rotates by repeating four stages; ATP-waiting dwell, rapid 80° substep rotation upon ATP binding, catalytic dwell in which ATP hydrolysis occurs, and rapid 40° substep rotation, probably upon the release of the last product.

ATP-driven rotation of F_0F_1 was demonstrated for *Propionigenium modestum* F_0F_1 with single-fluorophore polarization (17), and for *Escherichia coli* F_0F_1 with disulfide cross-linking (18, 19), direct visualization (20), and fluorescence resonance energy transfer (21). In general, ATP-driven rotation of F_0F_1 can differ from that of F_1 because proton transport through F_0 and interaction between the c -subunit ring (rotor) and ab_2 (stator) during rotation may modify the rotation. It was reported that the central rotor rotates counterclockwise when viewed from membrane side, and there are three pauses (each 19–30 ms) in one revolution, likely corresponding to the catalytic dwell (21, 22). The rotation of F_0F_1 driven by proton flow was also observed, and the direction of the rotation is opposite of that of the ATP-driven rotation (22). However, in general, knowledge on rotation of F_0F_1 has yet been very limited. For example, even the following basic motor natures of F_0F_1 remain unknown: (i) whether ATP-binding dwell appears at low [ATP], (ii) how the rotation depends on [ATP], (iii) whether the observed rotation consists of the sequence of 80° and 40° substep rotation or other new substep(s) exists at different angular position(s), (iv) whether catalytic dwell is composed of two events, as observed for F_1 , and (v) how rotation changes when a reversible F_0 inhibitor is present. To address these questions, we isolated thermophilic F_0F_1 , which kept structural integrity in a detergent, immobilized it on a glass surface through the β -subunits, attached a small (80 nm) bead to the c -subunit ring as a rotation probe whose viscous friction was low enough to allow full-speed rotation, and observed ATP-driven rotation with a fast camera. The results reveal that basic natures of ATP-driven rotation of

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Abbreviations: ATP γ S, adenosine 5'-[γ -thio]triphosphate; DCCD, N,N' -dicyclohexylcarbodiimide; LPC, lysophosphatidylcholine; Ni^{2+} -NTA, Ni^{2+} -nitrilotriacetic acid; rps, revolutions per second; TBT-Cl, tributyltin chloride.

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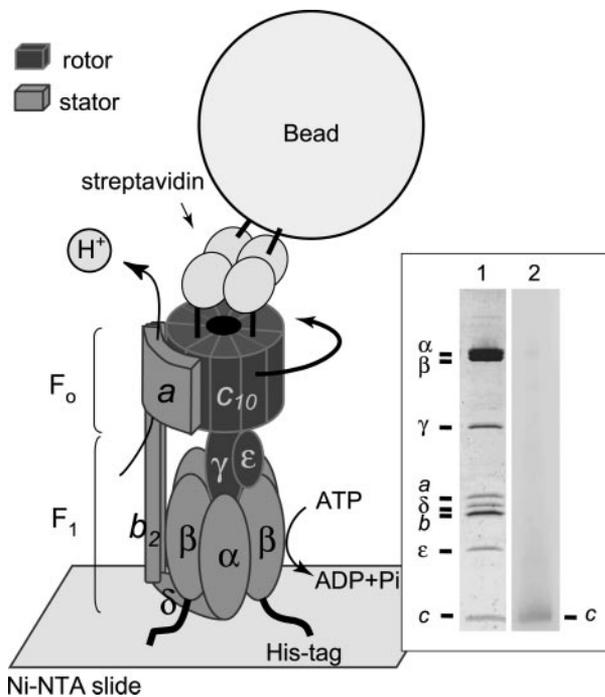


Fig. 1. Experimental system. ATP-driven rotation of the bead attached to the c-subunit ring in F_0F_1 was observed under dark-field microscopy and recorded with a fast camera. (Inset) F_0F_1 from thermophilic *Bacillus* PS3 used for observation of rotation. Lane 1, PAGE in the presence of sodium dodecyl sulfate of F_0F_1 purified in LPC and stained by Coomassie brilliant Blue; lane 2, immunoblots stained by alkaline phosphatase-streptavidin conjugates that show specific biotin labeling of the c-subunit.

F_0F_1 are almost unaltered from those of F_1 , except for the response to an F_0 -specific inhibitor.

Materials and Methods

Isolation of F_0F_1 . A plasmid pTR19-ASDS-CNCR3 for mutant F_0F_1 (cSer2Cys/ β -His₁₀ tags) was made by the Mega-primer method (23) using a plasmid, pTR19-ASDS (24), an expression vector for F_0F_1 complex of thermophilic *Bacillus* PS3. The mutant F_0F_1 was expressed constitutively in *E. coli* strain DK8 [Δ (*uncB-uncC*), *ilv::Tn10*] that lacks whole F_0F_1 genes, and the inverted membrane vesicles were prepared by the procedures described (24). After washing a 1-ml suspension of the vesicles (20 mg of protein per ml) with 5-fold volume of buffer PA3 (10 mM HEPES/KOH, pH 7.5/5 mM MgCl₂/10% glycerol) containing 2% sodium cholate by centrifugation, the vesicles were dissolved in 2 ml of buffer PA3 containing 2% (wt/vol) octaethylene glycol monododecyl ether and 100 μ M Tris (2-carboxyethyl) phosphine, and incubated for 30 min at room temperature. After adjusting the pH of the solution to 7.0 by phosphate buffer, 6-*N'*-[2-(*N*-maleimido)ethyl]-*N*-piperazinyl-amidohexyl-D-biotinamide (Dojindo) was added (50 μ M), and incubated for 30 min at room temperature. Biotinylation was quenched with 100 μ M DTT, and the suspension was diluted 6-fold with buffer M (20 mM potassium phosphate buffer, pH 7.5/100 mM KCl) containing 0.1% lysophosphatidylcholine (LPC) and 20 mM imidazole. This suspension was applied to a Ni²⁺-nitrilotriacetic acid (Ni²⁺-NTA) column (Qiagen, Valencia, CA) equilibrated with the same buffer. After washing with 10 volumes of the equilibration buffer, the protein was eluted with buffer M containing 0.1% LPC and 200 mM imidazole. The eluate containing F_0F_1 was applied to a Soft-Link avidin column (Promega) equilibrated with buffer M containing 0.05% LPC. The column was washed with 10 volumes of the same buffer, and

the protein was eluted with buffer M containing 0.05% LPC and 10 mM D-biotin. The purification was finished within 4 h, and the purified sample was used within 1 day. The purified F_0F_1 contained 0.9 mol of ADP and 0.8 mol of ATP per mol of F_0F_1 as endogenously bound nucleotides. Specific biotinylation of the F_0c -subunit was confirmed by immunoblotting using streptavidin-alkaline phosphatase conjugates (Promega).

Beads. Colloidal gold (diameter of 40 or 80 nm; British BioCell International) was incubated in 100 mM borate buffer, pH 8.2, containing 20 mg/ml BSA for 16 h at 37°C. BSA-coated colloidal gold was precipitated by centrifugation, and resuspended in 100 mM borate buffer, pH 8.2, with 10 mg/ml BSA. The suspension was biotinylated with 0.4 mg/ml 15-[(biotinoyl)amino]-4,7,10,13-tetraoxapentadecanoic acid, *N*-hydroxysuccinimidylester (Pierce) for 3 h at room temperature. After removing unreacted biotin by centrifugation, 1 mg/ml streptavidin was added. After the excess streptavidin was removed by centrifugation, the pellet was resuspended in 2 mM potassium phosphate, pH 7.0, containing 0.05% polyethylene glycol and stored at 4°C. In the experiments to test the inhibition of rotation by *N,N'*-dicyclohexylcarbodiimide (DCCD), larger beads (streptavidin-coated microspheres, 0.56 μ m, Bangs Laboratories, Carmel, IN) were used.

Rotation Assay. A flow cell was made of a Ni²⁺-NTA-coated cover glass and a slide glass separated by two spacers with 50- μ m thickness. At first, buffer R (50 mM HEPES/KOH, pH 7.5/100 mM KCl/10 mg/ml BSA/0.05% LPC/20 mM imidazole) was infused into the flow cell and incubated for 5 min to block nonspecific binding of the enzyme. Biotinylated His₁₀-tagged F_0F_1 (1–5 nM) in buffer R was infused. After 10 min, unbound F_0F_1 was washed with buffer R. Then, streptavidin-coated beads (2×10^9 particles per milliliter) in buffer R1 (50 mM HEPES/KOH, pH 7.5/100 mM KCl/10 mg/ml BSA/0.05% LPC) were infused, and incubated for 10 min. Unbound beads were removed with buffer R1 containing 5 mM MgCl₂, an ATP-regeneration system, and ATP at indicated concentrations, and observation of rotation was started. When indicated, ATP- γ S was added instead of ATP, and an ATP-regeneration system was omitted in this case. The number of the beads at the glass surface depended on the concentrations of biotinylated F_0F_1 infused to the flow cell. When nonbiotinylated enzyme was infused, it was almost the same as the case when no protein was infused, and we could not find any rotating beads in this case. Inhibitory effect of DCCD was assessed by observing the rotation of F_0F_1 incubated with 50 μ M DCCD for 30 min at room temperature before infusion into the flow cell. Beads were observed with a dark-field microscopy (IX-70, Olympus) with a $\times 100$ objective lens (numerical aperture of 1.35, Olympus) and a dark-field condenser (numerical aperture of 1.2–1.4, Olympus). Both the objective lens and the condenser were warmed by lens (condenser) heater (Tokai Hit) to maintain temperature of the flow cell at 37°C. Bead images were recorded as an eight-bit AVI file with a fast-framing charge-coupled device camera (Hi-Dcam, NAC Image Technology) at the indicated frame rate. To analyze the acquired image data, custom software (created by R. Yasuda; Cold Spring Harbor Laboratory, Cold Spring Harbor, NY) was used. Rotation rate was obtained from the average of 20 continuous revolutions without unnatural interruption.

Other Assays. ATPase activity was measured at 37°C with an ATP regeneration system (25). The assay solution was composed of buffer R1 containing 5 mM MgCl₂, 1 mM ATP-Na, 2 mM phosphoenolpyruvate, 100 μ g/ml lactate dehydrogenase, 100 μ g/ml pyruvate kinase, and 0.2 mM NADH. DCCD inhibition of ATP hydrolyzing activity was measured as described for the inverted membrane vesicles (24), or after a 30-min preincubation

Table 1. ATPase activity sensitivity of F_oF₁ to DCCD inhibition

Samples	Residual activity +DCCD, %
Membrane vesicles	25
Reconstituted vesicles	20
F _o F ₁	15

The samples were pretreated with 50 μM DCCD for 30 min and subjected to the assays. Membrane vesicles were prepared from *Escherichia coli* cells expressing F_oF₁ from thermophilic *Bacillus* PS3. Purified F_oF₁ was incorporated into reconstituted vesicles. ATPase activity was measured in the same solution used for rotation observation. Other experimental details are described in *Materials and Methods*.

with DCCD (50 μM) for the purified enzymes. To measure inhibitory effect of tributyltin chloride (TBT-Cl) on ATP hydrolyzing activity, indicated concentrations of TBT-Cl was added to the assay solution before the start of the reaction. Because the inhibitory effect of TBT-Cl on ATPase activity, as well as on rotation, tended to be relieved gradually as time passed (for an unknown reason), we collected rotation data quickly after the initiation of measurements. Reconstitution of F_oF₁ into liposomes was performed by the method described (24). Protein concentrations were determined by the BCA protein assay kit (Pierce) with BSA as a standard.

Results

Intact F_oF₁ in Detergent. In this study, we used F_oF₁ of thermophilic *Bacillus* PS3 expressed in the plasma membranes of an F_oF₁-deficient *E. coli* strain, DK8. A detergent octaethylene glycol monododecyl ether was used to solubilize F_oF₁ from the membrane vesicles and it was substituted with LPC at the next step of purification. We have tested various detergents, but this combination of detergents gave the most efficient solubilization, and the F_oF₁ preparation had the most stable, intact coupling properties. The F_oF₁ has His₁₀ tags at the N terminus of the β-subunits to immobilize onto a glass surface and has a cysteine residue at the second position from N terminus of the c-subunit for biotinylation to attach the beads. The enzyme was purified with a Ni²⁺-NTA column and a Soft-Link avidin column (Fig. 1 *Inset*, lane 1). These procedures should help to remove free F₁ and F_o, if any. Specific biotinylation of the c-subunit was confirmed by immunoblotting (Fig. 1 *Inset*, lane 2). The purified F_oF₁ in LPC comprised eight kinds of subunits and exhibited DCCD-sensitive ATPase activity; ≈85% of the activity was inhibited by DCCD in the solution used for observation of rotation (Tables 1 and 2). This degree of inhibition is similar to that of the intact F_oF₁ embedded in membranes, either in the membrane vesicles, or in the reconstituted vesicles. It has been known that DCCD covalently labels an essential carboxyl residue of the c-subunit, blocks proton transport, and consequently, if

Table 2. Effect of DCCD on rotation of F_oF₁

Trial	No. of rotating beads	
	+DCCD	-DCCD
1	2	10
2	1	10
3	0	9
4	2	12
5	1	11
6	1	8
Total	7	60

Rotating beads were looked for in 120 optical fields of a unit area (~45 × 55 μm²) in 10 min in one trial. Other experimental details are described in *Materials and Methods*.

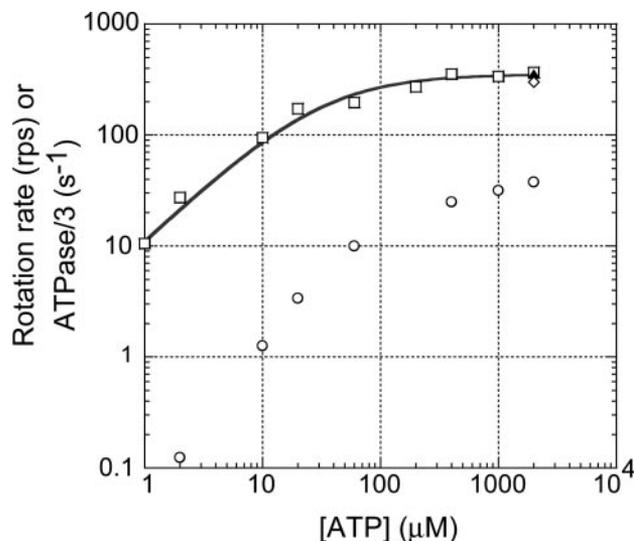


Fig. 2. Dependency of rotation rate and bulk-phase ATPase of F_oF₁ on [ATP]. □, rotation observed with an 80 nm-bead; ◇, rotation observed with a 40 nm-bead; ▲ rotation of F₁ observed with an 80 nm-bead; ○, bulk-phase ATPase activities measured in the rotation buffer. The line shows fit with Michaelis–Menten kinetics, $V = V_{max}[ATP]/(K_m + [ATP])$, where $V_{max} = 352 \pm 16$ rps, and $K_m = 31 \pm 7$ μM. Values are means ± SE.

F_oF₁ is intact, prevents ATP hydrolysis/synthesis. Unlike F_oF₁, ATPase activity of the isolated F₁ was not affected by DCCD at pH values of >7.5. (data not shown). Therefore, high sensitivity to DCCD inhibition is a good indication of the intactness of our F_oF₁ preparation. The reconstituted vesicles containing purified F_oF₁ showed substantial ATP-driven proton-pumping activity, comparable with that of the authentic wild-type thermophilic F_oF₁ (24). Based on these observations, we concluded that the purified F_oF₁ in LPC had intact ATPase activity that coupled proton transport. This conclusion was further confirmed by TBT-Cl sensitivity of F_oF₁ as described later.

V_{max} Rotation of F_oF₁. F_oF₁ was immobilized on a glass surface through β-subunits, and an 80-nm bead was attached to the c-subunit ring. Because the size of the bead was much larger than the c-subunit ring (≈5 nm) (26), a single bead could be tethered to several c-subunits in the same ring, although two beads could not attach to the same c-subunit ring. ATP-driven rotation of the bead attached obliquely to the c-subunit ring was observed at 37°C under dark-field microscopy equipped with a fast camera (Fig. 1). Functional integrity of immobilized F_oF₁ was confirmed by the observations that the number of rotating beads was drastically decreased to 12% (7:60) by DCCD-pretreatment of F_oF₁ (Table 2). In the case of rotation of F₁, on the contrary, pretreatment of F₁ with DCCD did not cause decrease in the number of the rotating molecules (data not shown). The direction of rotation was exclusively viewed counter clockwise from the membrane side. We observed the rotation at various [ATP] and found that rotation rates obeyed a simple Michaelis–Menten kinetics with a V_{max} of 352 ± 16 revolutions per second (rps) and a K_m of 31 ± 7 μM (Fig. 2, □). We used an 80-nm single bead as a rotation probe, but the rotation rate observed with a 40-nm single bead at 2 mM ATP (≈300 rps, Fig. 2, ◇) was almost the same as that of an 80-nm bead, indicating that the viscous friction in the rotating 80-nm bead did not slow down the rotation rate of F_oF₁. It is noteworthy that the rotation rate of F_oF₁ under V_{max} conditions was very similar to the rotation rate of F₁ (≈320 rps, Fig. 2, ▲) under the same conditions. Therefore, any events occurring at the F_o portion during rotation, such as the interaction between rotor (c-subunit ring) and stator (ab₂-subunits)

and proton transport through F_o , does not limit rotation rate of F_oF_1 , at least in the absence of the electrochemical gradient of protons.

Rotation Rate Versus ATPase Activity. The bulk-phase steady-state ATPase activities of the purified F_oF_1 in the same solution used for observation of rotation were only $\approx 10\%$ at 2 mM ATP, and even $<1\%$ at 2 μM ATP of the expected values from the rotation rates (Fig. 2, \circ). This result indicates that $>90\%$ population of F_oF_1 molecules are not working at a given moment, or, in other words, a single molecule spends $\approx 90\%$ of time in inactive state(s). Indeed, we noticed that rotating molecules usually stopped rotation after several seconds of continuous rotation. Sometimes, the same molecule resumed rotation after a while. We also often saw, under the microscopic field, that a previously nonrotating bead started rotating. The real reason why such high fractions of F_oF_1 molecules are in inactive state(s) is not known, but it should be noted that F_1 ($\alpha_3\beta_3\gamma$ subcomplex) also shows similar large discrepancy of rates between rotation and bulk-phase steady-state ATPase activity. In the case of F_1 , the responsibility for the discrepancy is thought to be the ADP-Mg inhibition, which is caused from the nonturnover retention of ADP-Mg at a catalytic site (27–29). The steady-state ATPase activity of F_1 is largely suppressed by the ADP-Mg inhibition, but, uninhibited activity that is almost consistent with the rotation rate, is estimated from the initial burst activity that appears upon initiation of ATPase assays (14). We speculate that ADP-Mg inhibition occurring in F_oF_1 can explain, at least partly, the discrepancy between rotation and bulk-phase ATPase activity, although we cannot estimate its quantitative contribution because F_oF_1 does not show the uninhibited initial-burst activity. Pronounced inhibition of ATPase activity of F_oF_1 at low [ATP] in Fig. 2 might be due to the inhibition by the endogenous inhibitor, ϵ -subunit, that exhibits inhibition at low [ATP] (30).

Stepwise Rotation. We analyzed the time course of rotation at 2 mM ATP (Fig. 3A). Under V_{max} conditions, it is expected that ATP binds to F_oF_1 very quickly, and the catalytic events occurring in the enzyme determine the rates of ATP hydrolysis and rotation. Therefore, the rotation would show only the catalytic dwell. Indeed, the histogram of angular distribution of centroid of bead (Fig. 3A Inset) indicated the presence of three favorable positions for a bead to make a brief pause. However, the expanded time course of rotation did not always show clear steps (see Fig. 6, which is published as supporting information on the PNAS web site), and we could not define each dwell time with certainty.

At 2 μM ATP, F_oF_1 rotated with discrete 120° steps (Fig. 3B), and the dwell time of the pauses between adjacent step-rotations was apparently dependent on [ATP], indicating that the observed pauses were the ATP-waiting dwells. In addition, very short dwells were often seen in the intermediate position in a 120° rotation. They are not obvious in the bead-centroid distribution plots (Fig. 3B Lower Inset) because these dwells would be buried behind spreading distribution of the long ATP-waiting dwells. Usually, the starting time points of the short dwells could be recognized in the rotation time course as an interruption of 120° rotation, but the end points were mostly unclear; the dwells transited to the next ATP-waiting dwell without showing discrete step rotation. Therefore, the total dwell time from the start of the short dwell to the end of the ATP-waiting dwell was analyzed. The histogram of the total dwells at 2 μM ATP showed a distinct peak, and was fitted by the sum of two exponential components that assumed two rate-limiting reactions (Fig. 3B Upper Inset). One of the time constants was [ATP]-dependent and corresponds to the ATP-waiting dwell. From the time constant, the ATP-binding rate ($k_{\text{on-ATP}}$) of F_oF_1 was estimated to be $3.6 \pm 0.1 \times 10^7 \text{ M}^{-1}\text{s}^{-1}$. This value is close to that of F_1 at 23°C ($3.0 \pm$

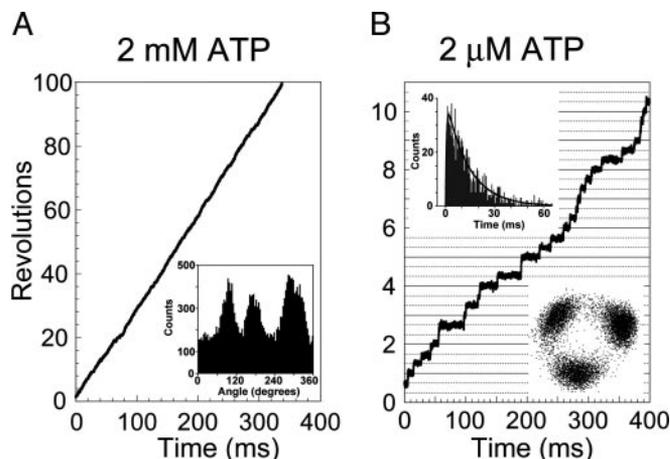


Fig. 3. Rotation of F_oF_1 driven by ATP. Bead images were recorded at 8,000 frames per second. (A) Time course of rotation at 2 mM ATP. (Inset) Histogram of angular distribution of the centroid of the bead image. (B) Time course of rotation at 2 μM ATP. (Upper Inset) Histogram of the total dwell times of rotation at 2 μM ATP. The total dwell time is defined as the time period from the start of the short dwell to the end of the ATP-waiting dwell. The black line is fit with the two rate constants, constant $\times [\exp(-k_1t) - \exp(-k_2t)]$, $k_1 = 0.073 \pm 0.002 \text{ ms}^{-1}$ (time constant = $13.8 \pm 0.4 \text{ ms}$) and $k_2 = 1.72 \pm 0.18 \text{ ms}^{-1}$ (time constant = $0.58 \pm 0.06 \text{ ms}$). k_1 is [ATP]-dependent and corresponds to the ATP-waiting dwell. k_2 is [ATP]-independent and corresponds to the catalytic dwell. Values are means \pm SE. Total counts of dwells are 1,079. Bin width is 0.5 ms. (Lower Inset) Trace of the centroid of the bead image.

$0.1 \times 10^7 \text{ M}^{-1}\text{s}^{-1}$) (14). The other time constant ($\approx 0.58 \text{ ms}$) was [ATP]-independent and corresponds to the catalytic dwell. This value is consistent with that ($\approx 0.95 \text{ ms}$) obtained from the V_{max} rotation.

From these results, we learned that F_oF_1 rotates by repeating ATP-waiting dwell and catalytic dwell. However, the very short lifetime of the catalytic dwell did not allow for its further analysis, and the angular position of the catalytic dwell and the number of events occurring in the catalytic dwell were not determined. To learn these answers, we extended the duration of the catalytic dwell by adopting a slowly hydrolyzable ATP analog, ATP γ S, as a substrate (15).

Rotation Driven by ATP γ S. F_oF_1 rotated in 1 mM ATP γ S at ≈ 20 rps, >10 times slower rate compared with the rate in 1 mM ATP (≈ 335 rps), and the discrete 120° steps were observed (Fig. 4A Lower Inset). The rotation in 100 μM ATP γ S was apparently very similar to that observed in 1 mM ATP γ S; rotation rate of ≈ 18 rps and 120° step rotation (data not shown). The dwell time between steps was not changed in two ATP γ S concentrations and should be the catalytic dwell. The histogram of the catalytic dwell of the rotation in 1 mM ATP γ S showed a peak, and was fitted with the sum of two exponential components that provided two time constants, ≈ 12.4 and $\approx 2.3 \text{ ms}$ (Fig. 4A Upper Inset). Taking the F_1 study on hydrolysis of ATP γ S as a reference (15), the long time constant is likely the one for cleavage of ATP γ S, and the short time constant for the release of the last product. The total of these time constants ($\approx 14.7 \text{ ms}$) agrees well with that ($\approx 16.7 \text{ ms}$) obtained from V_{max} (≈ 20 rps). In 10 μM ATP γ S, the 120° step was further split into two substeps (Fig. 4B Lower Inset). The angles of the two substep rotations were roughly 80° and 40° that are the same as observed for F_1 . The pauses between adjacent 40° and 80° substep rotations depended on ATP γ S concentrations, and, hence, the ATP γ S-waiting dwell. The histogram of the ATP γ S-waiting dwell was fitted well with a single-exponential component (Fig. 4B Upper Inset), and the ATP γ S-binding rate ($k_{\text{on-ATP}\gamma\text{S}}$) of F_oF_1 was estimated to be

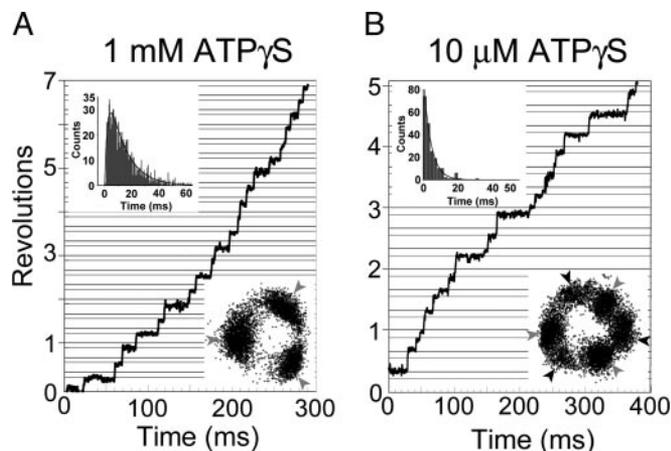


Fig. 4. Rotation of F_0F_1 driven by $ATP\gamma S$. Bead images were recorded at 2,000 frames per second. Gray horizontal lines are placed 40° below black lines. (A) Time course of the rotation of F_0F_1 at 1 mM $ATP\gamma S$. (Upper Inset) Histogram of the pauses before 40° substeps (catalytic dwell) in rotation at 1 mM $ATP\gamma S$. The line is fit with the two rate constants, constant $\times [\exp(-k_1t) - \exp(-k_2t)]$, $k_1 = 80.8 \pm 3.5 \text{ s}^{-1}$ (time constant = $12.4 \pm 0.5 \text{ ms}$) and $k_2 = 445 \pm 36 \text{ s}^{-1}$ (time constant = $2.25 \pm 0.18 \text{ ms}$). Values are means \pm SE. Total counts of dwells are 1,028. Bin width is 0.5 ms. (Lower Inset) Trace of the centroid of the bead image. Gray arrows indicate the positions of catalytic dwell. (B) Time course of the rotation of F_0F_1 at $10 \mu\text{M } ATP\gamma S$. (Upper Inset) Histogram of the pauses before 80° substep rotations ($ATP\gamma S$ -binding dwell) in rotation at $10 \mu\text{M } ATP\gamma S$. The line is a single-exponential fit, constant $\times \exp(-kt)$ where $k = 238.3 \pm 5.3 \text{ s}^{-1}$ (time constant = $4.2 \pm 0.1 \text{ ms}$). Values are means \pm SE. Total counts of dwells are 202. Bin width is 2 ms. (Lower Inset) Trace of the centroid of the bead image. Black arrows indicate the positions of $ATP\gamma S$ -binding dwell.

$2.4 \pm 0.1 \times 10^7 \text{ M}^{-1}\text{s}^{-1}$. This value is in the same order of $k_{\text{on-ATP}}$ ($3.6 \pm 0.1 \times 10^7 \text{ M}^{-1}\text{s}^{-1}$) of F_0F_1 , and $k_{\text{on-ATP}\gamma S}$ ($2.6 \pm 0.1 \times 10^7 \text{ M}^{-1}\text{s}^{-1}$) of F_1 (15). The pauses between adjacent 80° and 40° substep rotations corresponded to the catalytic dwell, and the analysis of the dwell-time histogram (not shown) gave two time constants, ≈ 17.3 and $\approx 2.4 \text{ ms}$, which is consistent with the values obtained from the rotation in 1 mM $ATP\gamma S$.

Effect of TBT-Cl on F_0F_1 Rotation. TBT-Cl has been known as an inhibitor of F_0F_1 (31, 32). It reacts noncovalently with the F_0 portion, probably with the F_0a -subunit, and prevents proton translocation-coupled ATP hydrolysis/synthesis (33). ATPase activity of our purified F_0F_1 measured in the same solution used for observation of rotation was efficiently ($\approx 80\%$) inhibited by $1 \mu\text{M}$ TBT-Cl (Fig. 5A, ●). No inhibition was observed in the presence of 0.3% lauryldodecylamine oxide (Fig. 5A, ○), a detergent known to disrupt stator/rotor interface of F_0 . ATPase activity of the isolated F_1 was also totally insensitive to TBT-Cl (Fig. 5A, □). Rotation of F_0F_1 in 1 mM ATP was slowed down significantly by TBT-Cl (Fig. 5B); the average rotation rate decreased to 4% ($\approx 15 \text{ rps}$) under the same condition was not affected by TBT-Cl (data not shown). We had expected to identify the angular position(s) where rotation would have paused in TBT-Cl, and 15 rps was slow enough to show any prolonged dwells. However, the rotating bead did not show any obvious pause at certain angular position (Fig. 5 Inset). We speculate that TBT-Cl interferes with the rotation of the F_0 portion at many angular positions, and makes the step-wise rotation obscure.

Discussion

In this study, we directly observed ATP-driven rotation of single-molecule F_0F_1 immobilized on a glass surface. To do so, F_0F_1 must be solubilized and purified without losing original

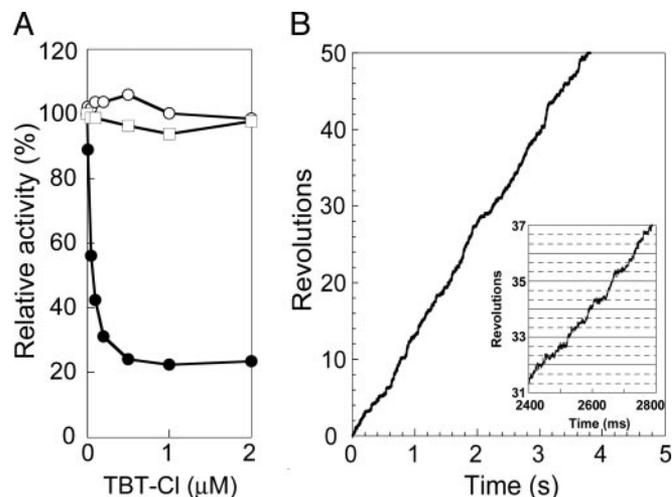


Fig. 5. Effect of TBT-Cl on ATPase activity of F_0F_1 (A) and rotation of F_0F_1 (B). (A) The ATPase activities of F_0F_1 were measured in the presence of indicated concentrations of TBT-Cl. Samples are F_0F_1 (●), F_0F_1 plus 0.3% lauryldodecylamine oxide (○), and F_1 (□). (B) Time course of the rotation of F_0F_1 in 1 mM ATP in the presence of $1 \mu\text{M}$ TBT-Cl. (Inset) Expanded time course of rotation.

structural integrity. It should be noted that the previous demonstrations of rotation of purified *E. coli* F_0F_1 immobilized on a glass surface were carried out by using the enzyme preparation, of which ATPase activity in the rotation buffer and the rotation itself were totally insensitive to DCCD (34, 35), and might have been the rotation of incomplete F_0F_1 , as stated in the subsequent paper (36). We have spent considerable effort to establish the procedures to isolate intact thermophilic F_0F_1 from the recombinant *E. coli* cells, and found that extraction from membranes by octaethylene glycol monododecyl ether and isolation in LPC gave the best yield and preparation. Both bulk-phase ATPase activity in the rotation buffer and the observed rotations under microscopy were efficiently inhibited by DCCD and TBT-Cl. Combination of this intact F_0F_1 and the submillisecond fast camera enabled us to learn several features of ATP-driven rotation of F_0F_1 .

Firstly, F_0F_1 rotates as fast as 350 rps (37°C). We observed rotations at 25°C and 45°C and obtained the V_{max} rotation rates of ≈ 230 and $\approx 650 \text{ rps}$, respectively. Therefore, extrapolated rotation rate at 60°C , an optimum growth temperature of *Bacillus* PS3, can reach $\approx 1,600 \text{ rps}$. Although reservation is needed concerning whether these enormous numbers are really the case, it is certain that the rotation rate of thermophilic F_0F_1 at 25°C is much faster than the rotation rate ($19 \text{ ms}/120^\circ$, that is, 17 rps, at 23°C) reported for *E. coli* F_0F_1 reconstituted into liposomes measured with fluorescence resonance energy transfer (22). However, in these measurements, uphill proton gradient should be immediately established by proton-pumping ATP hydrolysis to reach an equilibrium, and the rate limiting can be the rate of passive proton leak through membranes. It is interesting to discover whether the rotation rate will increase by addition of an uncoupler. One might argue whether fast rotation of $>200 \text{ rps}$ at V_{max} is common in F_0F_1 (and F_1) from various sources. Some reported values of very high ATPase activities would predict rapid rotations such as bovine mitochondrial F_1 ($\approx 310 \text{ rps}$) (37), yeast mitochondrial F_1 ($\approx 280 \text{ rps}$) (38), and *E. coli* F_0F_1 ($\approx 300 \text{ rps}$) (39). Much lower ATPase activities corresponding to $10 \approx 100 \text{ rps}$ were also reported in many papers. However, if a significant fraction of molecules in the bulk solution are in the ADP-Mg-inhibited state or other inactive states, as in the case of thermophilic F_0F_1 , real ATPase activity specific for the working enzymes should be higher, and the

